T 64 (6) V 64 (6)

ИЗВЕСТИЯ ВЫСШИХ УЧЕБНЫХ ЗАВЕДЕНИЙ. Серия **«ХИМИЯ И ХИМИЧЕСКАЯ ТЕХНОЛОГИЯ»**

ChemChemTech

2021

2021

DOI: 10.6060/ivkkt.20216406.6377

УДК: 537.525

ПАРАМЕТРЫ ГАЗОВОЙ ФАЗЫ И РЕЖИМЫ РЕАКТИВНО-ИОННОГО ТРАВЛЕНИЯ Si И SiO₂ В БИНАРНЫХ СМЕСЯХ Ar + CF4/C4F8

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Проведено сравнительное исследование электрофизических параметров плазмы, стационарного состава газовой фазы и закономерностей реактивно-ионного травления Si и SiO₂ в плазме бинарных смесей CF₄ + Ar и C₄F₈ + Ar в условиях индукционного BH 13,56 МГц разряда. Фиксированными параметрами процесса являлись давление плазмообразующего газа (6 мтор) и уровни мощности, подводимой от источников возбуждения плазмы (700 Вт) и смещения (200 Вт). Схема исследования включала диагностику плазмы зондами Лангмюра и О-мерное (глобальное) моделирование плазмохимических процессов. Показано, что исследованные смеси проявляют близкие свойства ионной компоненты плазмы и электронного газа, но характеризуются существенными отличиями в кинетике атомов и радикалов. Особенностями системы C4F8 + Ar в исследованном диапазоне условий являются более высокие концентрации полимеробразующих радикалов, меньшие концентрации атомов фтора и более слабая зависимость их концентрации от доли Аг в смеси. При исследовании кинетики травления установлено, что а) увеличение доли аргона в смесях CF4 + Ar и C₄F₈ + Ar сопровождается качественно различным изменением скоростей травления Si и SiO2; и б) наблюдаемые зависимости скорости от доли Ar в обеих смесях не согласуются с поведением плотности потока атомов фтора. Очевидно, что такая ситуация соответствует изменению вероятности взаимодействия атомов с обрабатываемой поверхностью. Предположено, что рост содержания Ar в смеси CF4 + Ar со слабой полимеризационной нагрузкой на обрабатываемую поверхность активирует гетерогенные химические реакции за счет интенсификации ионно-стимулированной десорбции продуктов травления и/или аморфизации поверхности. Подобный эффект в смеси C4F8 + Ar с высокой полимеризационной нагрузкой может быть отнесен к снижению толщины фторуглеродной полимерной пленки, что облегчает доступ атомов F к обрабатываемой поверхности.

Ключевые слова: Si, SiO₂, травление, полимеризация, поток атомов фтора, поток энергии ионов, эффективная вероятность взаимодействия

Для цитирования:

Ефремов А.М., Бетелин В.Б., Медников К.А., Кwon К.-Н. Параметры газовой фазы и режимы реактивно-ионного травления Si и SiO₂ в бинарных смесях Ar + CF₄/C₄F₈. *Изв. вузов. Химия и хим. технология*. 2021. Т. 64. Вып. 6. С. 25–34

For citation:

Efremov A.M., Betelin V.B., Mednikov K.A., Kwon K.-H. Gas-phase parameters and reactive-ion etching regimes for Si and SiO₂ in binary Ar + CF_4/C_4F_8 mixtures. *ChemChemTech* [*Izv. Vyssh. Uchebn. Zaved. Khim. Khim. Tekhnol.*]. 2021. V. 64. N 6. P. 25–34

GAS-PHASE PARAMETERS AND REACTIVE-ION ETCHING REGIMES FOR Si AND SiO₂ IN BINARY Ar + CF4/C4F8 MIXTURES

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The comparative study of plasma electro-physical parameters, steady-state gas phase compositions and reactive-ion etching kinetics for Si and SiO₂ in binary $CF_4 + Ar$ and $C_4F_8 + Ar$ gas mixtures were studied under conditions of 13.56 MHz inductive RF discharge. As fixed input parameters, we used the total pressure of feed gas (6 mTorr) as well as power levels supplied by plasma excitation source (700 W) and bias source (200 W). The investigation approach combined plasma diagnostics experiments with double Langmuir probe and 0-dimensional (global) model for the chemistry of neutral species. It was shown that investigated gas mixtures exhibit quite close properties in respect to both ions-related parameters and electron gas while are characterized by sufficient differences in kinetics of atoms and radicals. The features of $C_4F_8 + Ar$ gas under the given set of processing conditions are the higher density of polymerizing radicals, the lower density of F atoms as well as the weaker sensitivity the last parameter to the change in Ar fraction in a feed gas. Etching experiments indicated that a) an increase in Ar fraction in $CF_4 + Ar$ and $C_4F_8 + Ar$ gas mixtures results in qualitatively different changes in Si and SiO₂ etching rates; and b) obtained dependencies of etching rates on Ar fraction in both gas mixtures contradict with the behavior of F atom flux. Obviously, such situation corresponds to the change in reaction probability of F atoms with the treated surface. It was suggested that an increase of Ar fraction in the low-polymerizing CF_4 + Ar plasma activates the heterogeneous chemical reaction through the intensification of ion-stimulated desorption of etching products and/or surface amorphization. The similar effect for the high-polymerizing $C_4F_8 + Ar$ plasma may be related to decreasing fluorocarbon film thickness that provides the better access of F atoms to the etched surface.

Key words: Si, SiO₂, etching, polymerization, fluorine atom flux, ion energy flux, effective reaction probability

INTRODUCTION

Until now, silicon-based electronic devices continue to play the leading role in various branches of industry and science. The chemical basement of such devices is the silicon itself (as the substrate material) as well as various functional over- and underlayers which build the device structure. Among those, silicon dioxide plays an important role being used as gate dielectric, spacer material and passivating coating [1-3]. As real devices always have the complicated multi-level architecture, their fabrication procedure mandatory includes the patterning (the dimensional etching) of various silicon-based materials (SiO₂, Si₃N₄, SiC) together with the silicon itself. In order to satisfy requirements to device dimension and performance, one mostly uses the reactive-ion etching (RIE) technique with fluorocarbon gas plasmas [2-4]. Such favorable combination provides the ability to adjust the output process characteristics (etching rate, anisotropy factor and etching selectivity) through contribu-

tions of chemical and physical etching pathways as well as through the balance between etching and surface polymerization. Accordingly, there were many experimental studies reported about RIE kinetics and mechanisms for Si and SiO₂ in various fluorocarbon gas systems (for example, Refs. [5-13]). When summarizing basic features of both etching and polymerization processes on the plasma-etched surface interface, following conclusions can be made:

- The chemical etching pathway under conditions of conventional RIE process (gas pressure below 50 mTorr, ion bombardment energy above 200 eV) is mainly provided by F atoms.

– The chemical interaction of F atoms with Si surface occurs spontaneously even at nearly room temperatures (due to the formation of high volatile SiF4) and exhibits the weak sensitivity to the ion bombardment with energies below $\sim 100 \text{ eV}$. Nevertheless, Si etching rate in high-polymerizing plasmas is controlled not only by the F atom density, but also depends on ion fluxes and ion energies. The latter reflects the contribution of the sputter etching as well as is caused by the change in polymer film thickness that influences the access of F atoms to the etched surface.

– The chemical interaction of F atoms with SiO₂ surface has the sufficient energy threshold (as the Si-O bond of ~ 799 kJ/mol is stronger than the Si-F one ~ 552 kJ/mol [14]) and cannot occur spontaneously at typical process temperatures. Due to this reason, the chemical etching of SiO₂ by F atoms requires preliminary Si-O bonds breaking by the ion bombardment and thus, is controlled by the ion flux at $\epsilon_i < 200$ eV. At the same time, the higher ion energies lead to the domination of the chemical etching pathway controlled by the F atom flux. At least, SiO₂ etching rate in the CF₄ + O₂ plasma shows the maximum at 30-40% O₂, as it was repeatedly mentioned for F atom density [15-17].

– An effective tool to adjust the balance between etching and polymerization effects is to mix a fluorocarbon gas with Ar and/or O_2 [1, 2, 18]. The fraction of additive component does influence kinetics of F atoms and polymerizing radicals in a gas phase as well as changes decomposition rates of fluorocarbon polymer films through both physical and chemical mechanisms. The steady-state thickness of polymer film on SiO₂ is lower than that on Si under one and the same processing conditions [5-7]. The reason is etching of polymer by oxygen atoms on the film/SiO₂ interface [18].

When analyzing above results, one can formulate at least two important problems which require more attention and research efforts. First, most of experimental works did not discuss both etching and polymerization effects with accounting for plasma parameters and plasma chemistry. As such, even very detailed and accurate studies [5-11] said nothing about mechanisms which transfer the change in operating conditions to heterogeneous process kinetics. And secondly, data for various gas systems and materials correspond to different processing conditions and reactor types. Accordingly, since those cannot be compared directly, some uncertainty in the selection of optimal gas chemistry for the purpose of given RIE process still does exist.

Our previous works suggested the advanced investigation scheme which combines etching experiments with plasma diagnostics by Langmuir probes and modeling of plasma chemistry [12, 13, 19-21]. Such an approach allowed one a) to understand how processing parameters do effect on electrons- and ions-related plasma characteristics; b) to determine key gas-phase reaction pathways influencing steadystate densities of F atoms and polymerizing radicals; and c) to analyze etching mechanisms with modelpredicted fluxes of plasma active species. The main idea of given study was to above research scheme to RIE process of Si and SiO₂ in CF₄ + Ar and C₄F₈ + Ar plasmas. Since corresponding fluorocarbon components are very different in their polymerizing abilities [18], the comparative study may clearly demonstrate differences in corresponding etching regimes.

EXPERIMENTAL AND MODELING DETAILS

Experimental setup and procedures

Both plasma diagnostics and etching experiments were performed in the planar inductively coupled plasma (ICP) reactor, the same as that used in our previous studies [12, 13]. Plasma was produced using the 13.56 MHz rf generator connected to the 5turn copper coil on the top side of reactor chamber. Another rf power source with 12.56 MHz biased the bottom electrode. The latter was to set the ion bombardment energy according to the negative dc bias voltage (-U_{dc}). Constant processing parameters were total gas flow rate (q = 40 sccm), gas pressure (p = 6mTorr), input power ($W_{inp} = 700$ W) and bias power $(W_{dc} = 200 \text{ W})$. As a variable value, we used the component mixing ratio in a feed gas. This parameter was adjusted by partial flow rates of individual gases (q_i) within the constant total gas flow rate. Therefore, the composition of gas mixture coming to the reactor chamber was characterized by component fractions of $y_i = q_i/q$.

Plasma diagnostics by the double Langmuir probe (DLP2000, Plasmart Inc.) delivered data on

electron temperature (T_e) and ion current density (J₊). The treatment of measured voltage-current curves followed for the well-known statements of the double Langmuir probe theory [22]. In order to minimize the contamination of probe tips by the fluorocarbon polymer, these were treated before and after each measurement in the 50% Ar + 50% O₂ plasma [19-21].

Etching kinetics for Si and SiO₂ were studied using fragments of non-oxidized or oxidized Si(100) wafer. Etched samples with a size of ~ 2×2 cm were placed on the bottom electrode and centered in radial direction. The bottom electrode had a built-in waterflow cooling system that allowed one to maintain its temperature (T_s) at the nearly constant value within the processing times $\tau \sim 5$ min. The small sample size allowed one to neglect the loading effect as well as to provide the etching regime controlled by heterogeneous process kinetics. Etched depths (Δh) were determined using the surface profiler (Alpha-Step 500, Tencor) for the processing time $\tau = 1$ min. For this purpose, we developed a partial surface masking by the photoresist AZ1512 with a thickness of $\sim 1.5 \,\mu\text{m}$. The quasi-linear shape of $\Delta h = f(\tau)$ curves in both gas systems pointed out on the steady-state etching regime. Accordingly, Si and SiO₂ etching rates were simply calculated as $R = \Delta h/\tau$.

Plasma modeling

Relationships between gas mixing ratios and steady-state densities of plasma active species were studied using a simplified 0-dimensional (global) model. Detailed information on model assumptions and algorithm may be found in Refs. [12, 13, 21]. Kinetic schemes (sets of reactions with related rate coefficients) were taken from our earlier works devoted to of $CF_4 + Ar/O_2$ [16, 23, 24] and $C_4F_8 + Ar/O_2$ [13, 20, 21] plasmas. These works have confirmed that a combination of given kinetic schemes with given modeling approach allows one to obtain the correct description of plasma chemistry in gas systems of our interest. As input model parameters, we used experimental data on T_e and J₊. The latter yielded the total density of positive ions n_{+} as well as the electron density n_{e} under an assumption of $n_e \approx n_+$. The neutral gas temperature (T_{gas}) was assumed to be ~ 600 K, as the typical value for close processing conditions, reactor type and geometry [12, 13]. As output parameters, we obtained steady-state densities of neutral species and their fluxes to the treated surface.

Phenomenological analysis of heterogeneous kinetics

The basic features of reactive-ion etching processes in fluorocarbon-based plasmas have been discussed in detail in Refs. [5-13, 19, 21, 28]. Accord-

ing to these works, basic approaches for the analysis of etching and polymerization kinetics may be summarized as follows:

– Under conditions of conventional RIE process, the measured etching rate R represents the combination of two summands, $R_{phys} + R_{chem}$. These correspond to sputter and chemical etching pathways, respectively.

– The rate of physical etching pathway R_{phys} (in our case – the sputtering of both target surface and fluorocarbon polymer film, the breaking of Si-O bonds and the desorption of low-volatile reaction products) is $Y_S\Gamma_+$, where Y_S is the ion-type-averaged process yield, and $\Gamma_+ \approx J_+/e$ is the ion flux. When taking in mind that $Y_S \sim (M_i\epsilon_i)^{1/2}$, the change in any physical etching pathway may be traced by the parameter $(M_i\epsilon_i)^{1/2}\Gamma_+$, characterizing the ion energy flux.

- The rate of chemical etching pathway R_{chem} (in our case - the interaction of F atoms with Si and SiO₂ surfaces) is $\gamma_R \Gamma_F$, where Γ_F is the flux of F atoms with the gas-phase density [F], and $\gamma_R \approx s_0(1-\theta)$ is the effective reaction probability. The last parameter exhibits the exponential-like dependence on the temperature of etched surface (through the sticking coefficient s₀, as follows from the Arrhenius law) as well as is sensitive to any plasma-related factor influencing the balance between free adsorption sites $(1 - \theta)$ and those passivated by reaction products and/or nonreactive species θ . In polymerizing plasmas, γ_R decreases with increasing thickness of the fluorocarbon polymer film when the latter reached the threshold thickness that provides $\Gamma_F'/\Gamma_F \ll 1$, where Γ_F' is the flux of F atoms on the polymer film/etched surface interface.

– The growth of polymer layer is provided by CF_x (x = 1, 2) radicals as well as appears to be slower in fluorine-rich plasmas. Therefore, the Γ_{pol}/Γ_F ratio, where Γ_{pol} is the total flux of polymerizing radicals, surely characterizes the polymer deposition rate, and the parameter $\Gamma_{pol}/(M_{i}\epsilon_{i})^{1/2}\Gamma_{+}\Gamma_{F}$ traces the change in the polymer film thickness.

RESULTS AND DISCUSSION

Correlations of processing conditions (gas pressure, input power and bias power) with electronsand ions-related plasma parameters in CF₄ and C₄F₈ gases have been studied in earlier works [24-27]. Due to this reason, below we will analyze only on differences between CF₄ + Ar and C₄F₈ + Ar plasmas under the condition of p, W_{inp} = const. According to Table, the main features may be formulated as follows:

– The electron temperature exhibits systematically higher values in C_4F_8 + Ar plasma as well as is characterized by somewhat different change vs. Ar

Изв. вузов. Химия и хим. технология. 2021. Т. 64. Вып. 6

content in a feed gas. The first phenomenon looks rather unexpectedly, as the C₄F₈ plasma contains bigger multi-atomic species (Fig. 1) with potentially more effective electron energy loss channels. At the same time, the total density of C₂F₄, C₂F₃ and CF₄ compounds in this gas system is lower compared with that for less-saturated CF_x (x \leq 3) radicals. As such, assuming that exactly these species provide the major electron energy consumption in C₄F₈ plasma, higher T_e values may be related to lower energy losses for vibrational and electronic excitations compared with those for CF4 molecules. An increase in Ar fraction in CF_4 + Ar mixture lowers electron energy loss for excitations of fluorocarbon species, but causes the nearly-proportional increase in the energy loss for ionization. The last effect is due to the condition $\varepsilon_1 k_1 < \varepsilon_2 k_1$ $\epsilon_2 k_2$, where R1: CF₄ + e \rightarrow CF₃⁺ + F + 2e ($\epsilon_1 = 15.9$ eV, $k_1 = 1.6 \times 10^{-9} \text{ cm}^3/\text{s}$ at $T_e = 3 \text{ eV}$) and R2: Ar + e \rightarrow Ar⁺ + 2e ($\epsilon_2 = 15.6 \text{ eV}$, $k_2 = 3.0 \times 10^{-9} \text{ cm}^3/\text{s}$ at $T_e = 3$ eV). That is why the change in CF₄/Ar mixing ratio at p, $W_{inp} = const results in T_e \approx const.$ At the same time, and increase of y_{Ar} in C_4F_8 + Ar plasma causes the much stronger increase in electron energy losses for ionization because of $\varepsilon_3 k_3 < \varepsilon_1 k_1$, where R3: CF₂ + e \rightarrow \rightarrow CF₂⁺ + 2e with $\varepsilon_3 = 10.0$ eV and k₃ = 2.1×10⁻⁹ cm³/s at $T_e = 3 \text{ eV}$. Such situation produces a decrease in T_e toward Ar-rich plasmas.

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Electrons- and ions-related plasma parameters in CF₄ + Ar and C₄F₈ + Ar plasmas *Таблица*. Параметры электронной и ионной

компонент в плазме CF4 + Ar and C4F8 + Ar									
y _{Ar} ,	$CF_4 + Ar$			$C_4F_8 + Ar$					
%	T _e , eV	n ₊ , cm ⁻³	-U _{dc} ,V	T _e , eV	n ₊ , cm ⁻³	-U _{dc} ,V			
0	3.6	$4.4 \cdot 10^{10}$	262	4.7	$3.9 \cdot 10^{10}$	278			
25	3.6	$4.5 \cdot 10^{10}$	249	4.5	$4.4 \cdot 10^{10}$	249			
75	3.8	$5.8 \cdot 10^{10}$	188	3.8	$8.3 \cdot 10^{10}$	198			

- The plasma density exhibits quite close absolute values in both gas systems as well as is characterized by similar changes vs. Ar content in a feed gas. A bit higher n_+ in pure CF₄ plasma at $y_{Ar} < 30\%$ is due to slower ion transport to chamber walls under the condition of close total ionization frequencies, as follows from $k_1 \approx k_3$. An increase in both n_+ and n_e toward Arrich plasmas is provided by the growth in total ionization frequencies. Such situation results from a combination of two conditions, such as a) the domination of k₂ over k₁ and k₃ due to higher ionization cross-section for Ar atoms; and b) the effective ionization of metastable atoms $Ar^{m} = Ar({}^{3}P_{0,1,2})$ through R4: $Ar^{m} + e \rightarrow$ \rightarrow Ar⁺ + 2e. The systematically higher ion current density (and thus, the ion flux) in the C_4F_8 + Ar plasma ($J_{+} = 1.05 - 2.36 \text{ mA/cm}^2$ for $C_4F_8 + Ar$ vs.

 $0.95 - 1.63 \text{ mA/cm}^2$ for CF₄ + Ar and at 0–75% Ar) is due to differences in masses and transport coefficients for dominant positive ions.

– The negative dc bias at W_{dc} = const exhibits slightly higher values in C₄F₈ + Ar plasma as well as always decrease with increasing y_{Ar}. That is why, there are no principal differences in both absolute values and tendencies for ion bombardment energies. Corresponding ranges are 284-210 eV for CF₄ + Ar and 308-220 eV for C₄F₈ + Ar at 0-75% Ar. As a results, the C₄F₈ + Ar plasma provides the systematically higher ion bombardment intensity, as can be seen from differences in (M_iε_i)^{1/2}Γ₊ (Fig. 2).

In the CF₄ + Ar plasma, dominant fluorinecontaining components are CF4 molecules together with CF₃ radicals and F atoms (Fig. 1(a)). The decreasing densities of CF_x radical toward lower "x" values $([CF_3] > [CF_2] > [CF] [24, 25])$ is due to their consequent formation mechanism in R5: $CF_x + e \rightarrow$ $CF_{x-1} + F + e$ as well as because of higher heterogeneous loss probabilities (sticking coefficients) for lower saturated CF_x species. The main formation pathways for fluorine atoms are R1 and R5 with x = 3and 4. The loss of fluorine atoms is mainly due to a couple of heterogeneous processes R6: $F + F \rightarrow F_2$ and R7: $F + CF_x \rightarrow CF_{x+1}$ while corresponding gasphase counterparts contribute the total recombination rate by less than 10%. The dilution of CF₄ by argon does not introduce new process for the production of F atoms, but influences the electron impact dissociation kinetics for CF_x species due to the simultaneous increase in Te and ne. In particular, one can obtain an increase dissociation frequencies (16.1-34.3 s⁻¹ for CF_4 in R1 + R5, 28.7-51.1 s⁻¹ for CF_3 in R5 and 40.1-74.2 s⁻¹ for CF in R5 at 0-75% Ar) as well as slower changes in densities of F atoms (by ~ 1.8 times at 0-75% Ar) and polymerizing radicals (by \sim 1.4 times for $[CF] + [CF_2]$ at 0-75% Ar).

In the C₄F₈ + Ar plasma, dominant F atom carriers are fluorocarbon components CF_x (x = 1, 2, 3) and C₂F_x (x = 3, 4) [22, 26, 27] (Fig. 1(b)). These are either first-step dissociation products appeared through R8: C₄F₈ + e \rightarrow 2C₂F₄ + e and R9: C₄F₈ + e \rightarrow \rightarrow C₃F₆ + CF₂ + e or fragments from the further decomposition of corresponding species in electronimpact processes. Among latters, the most important are R5 for x = 2, R10: C₃F₆ + e \rightarrow C₂F₄ + CF₂ + e, R11: C₂F₄ + e \rightarrow 2CF₂ + e and R12: C₂F₄ + e \rightarrow C₂F₃ + +F + 2e. The main source of F atoms is represented by R5 for x = 1, 2, 3 while their decay is noticeably contributed by R13: C₂F₄ + F \rightarrow CF₂ + CF₃ (k₁₃ \sim 4·10⁻¹¹ cm³/s). An increase in y_{Ar} does not affect the efficiency of electron-impact reactions (due to opposite changes in T_e and n_e) and thus, causes the nearly proportional decrease in densities of fluorocarbon radicals (for example, by ~ 3.8 times for [CF] + [CF₂] at 0-75% Ar). At the same time, the effect of y_{Ar} on the F atom density is much weaker (by ~ 1.7 times at 0-75% Ar) and is comparable with that in CF₄ + Ar plasma. The last feature is not connected with electron-impact kinetics, but comes from the rapidly decreasing F atom decay frequency in the gas-phase reaction R13.

Above data allow one to conclude that the $C_{4}F_{8}$ + Ar plasma provides the much higher density of polymerizing radicals (mostly due to higher formation rates of CF₂ in R9, R11 and R13) while demonstrates only a bit lower density of F atoms (mostly due to their higher decay frequency in R13). Accordingly, the difference between fluxes of polymerizing radicals and polymer deposition rates in these gas systems reaches two orders of magnitude (Figs. 2(a, b)). The similar gap is also between steady-state thicknesses of polymer films, as can be seen from Fig. 2(d). The last conclusion is in the evident agreement with F/C ratios for corresponding fluorocarbon molecules.



Fig. 1. Steady-state densities of neutral species in CF₄ + Ar (a) and C₄F₈ + Ar (b) plasmas. Black bars correspond to pure fluorocarbon gas while grey bars are for mixtures with 50% Ar Рис. 1. Стационарные концентрации нейтральных частиц в плазме CF₄ + Ar (a) and C₄F₈ + Ar (б). Черные столбцы относятся к чистому фторуглеродному газу, а серые - к смесям, содержащим 50% Ar

Important features of the CF₄ + Ar plasma are nearly constant values of Γ_{pol}/Γ_F and $\Gamma_{pol}/(M_{i\epsilon_i})^{1/2}\Gamma_+\Gamma_F$ under the condition of $y_{Ar} = 0.75\%$. In fact, this points out on the weak effect of CF₄/Ar mixing ratio on both polymer deposition rate and its residual amount on the plasma-treated surface. Similar results were obtained in Ref. [5] by experiments. Oppositely, an increase of y_{Ar} in the C₄F₈ + Ar plasma leads to the nearly proportional (by ~ 3.8 times at 0-75% Ar) decrease in the parameter $\Gamma_{pol}/(M_{i\epsilon i})^{1/2}\Gamma_{+}\Gamma_{F}$ and thus, causes the similar change in the polymer film thickness. As such, the variation of C₄F₈/Ar mixing ratio provides a real tool to adjust both polymer film thickness and related output characteristics of RIE process.

In etching experiments, it was found that etching rates of Si and SiO₂ in pure Ar plasma (in fact, rates of physical sputtering, R_{phys}) are quite close and do not exceed 10 nm/min. Obviously, corresponding values are much lower compared with those measured in fluorocarbon-containing gas mixtures, as shown in Fig. 3. When analyzing these data together with fluxes of plasma active species, some suggestions about etching mechanisms and related effects can be made. First, the condition $R_{phys} \ll R$ means that the dominant etching pathway in all cases is the ion-assisted chemical reaction. Second, identical shapes of $R = f(y_{Ar})$ curves for Si and SiO₂ for each gas system probably reflect the situation that corresponding etching processes are provided by same active species and

> have same limiting stages. At least, this suggestion is supported by the experimental study made by Standaert et. al [5]. And thirdly, the evident dissimilarity in etching rate behaviors in Figs. 3(a) and 3(c) point out on different etching regimes in CF₄- and C₄F₈based plasmas. According to previous etching experience, one can suggest that the lowpolymerizing CF_4 + Ar plasma is featured by the ion-assisted chemical reaction under the condition of thin or even non-continuous polymer film. Oppositely, the highpolymerizing C_4F_8 + Ar plasma provides the etching regime controlled by the polymer thickness through the transport of etchant species to the film/etched surface interface.

In order to verify above suggestions concerning etching regimes, it is necessary to analyze correlations between R_{chem} and fluxes of plasma active species. In particular, non-monotonic Si and SiO₂ etching rates from Fig. 3(a) contradict with monotonic changes in both $(M_i \varepsilon_i)^{1/2} \Gamma_+$ and Γ_F as well as correspond to monotonically increasing effective reaction probabilities $\gamma_R = R_{chem}/\Gamma_F$ (Fig. 3b), where $R_{chem} = R - R_{phys}$. Obviously, an increase in γ_R with increasing Ar fraction in a feed gas cannot be associated with the polymer film thickness, as follows from very weak changes of both Γ_{pol}/Γ_F and $\Gamma_{pol}/(M_i\epsilon_i)^{1/2}\Gamma_+\Gamma_F$ in the range of 0-75% Ar. At the same time, the good correlation between γ_R and $(M_i \epsilon_i)^{1/2} \Gamma_+$ (Fig. 2c) allows one to assume that the ion bombardment activates heterogeneous chemical reaction through both desorption of reaction products and formation of adsorption sited for F atoms. In fact, this is a typical situation for the ion-assisted chemical reaction in non-polymerizing plasmas [18, 29, 30]. In addition, similar etching behaviors for Si and SiO₂ surely mean that the SiO₂ etching rate is not limited by the production of adsorption sites for F atoms in the ion-driven process $SiO_x(s.) \rightarrow Si(s.) + xO$. Probably, such situation results from the high ion energy flux that provides the fast destruction of oxide bonds on the plasma-etches surface interface.

Monotonically increasing Si and SiO₂ etching rates from Fig. 3(c) also contradict with the change of Γ_F and also correspond to increasing effective reaction probabilities (Fig. 3d). The main difference with a previous case is that the change of γ_R not only follow the tendency of $(M_{i\epsilon i})^{1/2}\Gamma_+$, but also demonstrates an agreement with a decrease in the polymer film thickness, as can be seen from Fig. 2(d). Taking into account high polymerizing ability of C₄F₈-based plasmas, one can reasonably suggest that a) the RIE etching process in pure C₄F₈ plasma is limited by the transport of etchant species through the thick polymer film; and b) the transition to Ar-rich plasmas activates chemical reaction by providing the better access of F atoms to the etched surface. Therefore, the formally similar changes in γ_R vs. y_{Ar} in CF₄ + Ar and C₄F₈ + Ar plasmas are caused by different mechanisms.



Fig. 2. Gas-phase-related parameters characterizing kinetics of etching and polymerization processes in CF₄ + Ar (1) and C₄F₈ + Ar (2) plasmas: a) total flux of polymerizing radicals; b) Γ_{pol}/Γ_F ratio characterizing the polymer deposition rate; c) parameter (M₁ ϵ_i)^{1/2} Γ_+ characterizing the polymer physical etching (sputtering) rate; and d) parameter $\Gamma_{pol}/(M_1\epsilon_i)^{1/2}\Gamma_+\Gamma_F$ which traces the polymer film thickness Puc. 2. Параметры газовой фазы, характеризующие кинетику травления и полимеризации в плазме CF₄ + Ar (1) и C₄F₈ + Ar (2): a) суммарный поток полимеробразующих радикалов; б) отношение Γ_{pol}/Γ_F , характеризующее скорость осаждения полимера; в) параметр (M₁ ϵ_i)^{1/2} Γ_+ , характеризующий скорость травления полимера; и с) параметр $\Gamma_{pol}/(M_1\epsilon_i)^{1/2}\Gamma_+\Gamma_F$ отслеживающий толщину полимерной пленки



Fig. 3. Etching rates (a, c) and effective reaction probabilities for F atoms (b, d) CF₄ + Ar (a, b) and C₄F₈ + Ar (c, d) plasmas Рис. 3. Скорости травления (a, b) и эффективные вероятности взаимодействия для атомов фтора (c, d) в плазме CF₄ + Ar (a, b) and C₄F₈ + Ar (c, d)

CONCLUSIONS

In this work, we discussed in a comparative scale reactive-ion etching kinetics and mechanisms for Si and SiO₂ in CF₄ + Ar and C₄F₈ + Ar plasmas.

It was shown that an increase in Ar faction in a feed gas always a) results in sufficient changes in plasma density (due to changes in total ionization rate); b) increases the intensity of ion bombardment; and c) causes the lower-than-proportional decrease in F atom density. The last phenomenon is due to either an increase in electron-impact dissociation frequencies for CF_x species (in the case of CF₄ + Ar plasma) or a decrease in F atom loss frequencies in gas-phase reactions (in the case of C₄F₈ + Ar plasma). Specific features of C₄F₈ + Ar plasma under the given set of processing conditions are the higher polymerization ability (as follows from lower F/C ration in the original fluorocarbon molecule) as well as the lower F atom density. It was found that the dominant etching mechanism in all cases is the chemical etching pathway while corresponding etching rates do not follow the behavior of F atom flux. This fact points out on the process-condition-dependent effective reaction probability for F atoms. The analysis of correlations between effective reaction probability and gasphase-related plasma parameters allowed one to suggest limiting stages and etching regimes. As a result, it was concluded that the formally similar changes of effective reaction probability for F atoms in CF₄ + Ar and C₄F₈ + Ar plasmas are due to different heterogeneous effects.

The research were carried out within the framework of the state assignment of the Federal State Institution «Scientific Research Institute for Sys-

Изв. вузов. Химия и хим. технология. 2021. Т. 64. Вып. 6

tem Analysis of the Russian Academy of Sciences» (fundamental research) on subject No. 0580-2021-0006 "Fundamental and applied research in the field of lithography limits in semiconductor technologies as well as physical and chemical etching processes for 3D nanometer dielectric structures for the development of critical technologies for the production of ECB. Investigations and developments of both models and constructions for microelectronic elements in the extended temperature range (from -60C to +300C)".

Работа выполнена в рамках государственного задания ФГУ ФНЦ НИИСИ РАН (проведение фундаментальных научных исследований) по теме № 0580-2021-0006 «Фундаментальные и прикладные исследования в области литографических пределов полупроводниковых технологий и физико-химических процессов травления 3D нанометровых диэлектрических структур для развития критических технологий производства ЭКБ. Исследование и построение моделей и конструкций элементов микроэлектроники в расширенном диапазоне температур (от -60С до +300С)».

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А.М. Ефремов и др.

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Поступила в редакцию 10.02.2021 Принята к опубликованию 01.04.2021

Received 10.02.2021 Accepted 01.04.2021